



Experiment 7:  
DENSITY



First Things First...  
Please Hand in Last Week's Lab!

# MASS

Mass is the number of protons and neutrons  
that an object has

# Mass vs. Weight

- MASS = amount of stuff or number of protons and neutrons
- Same everywhere in universe
- WEIGHT = gravitational attraction on a given mass
- Different depending on where you are, i.e. you weigh less on the moon

# Volume

The VOLUME of an object is simply how much space it takes up



Aggie Stadium –  
LOTS of volume!



Mini-Me: Not so  
much...

# Which Has More Mass?



A pound of feathers? Or a pound of lead?

# Trick Question!!

- They both have the same mass... 1 lb
- In other words, they both have the same amount of stuff – protons and neutrons

BUT... how many feathers do you need to get 1 lb vs. how much lead to get 1 lb?

# DENSITY!!

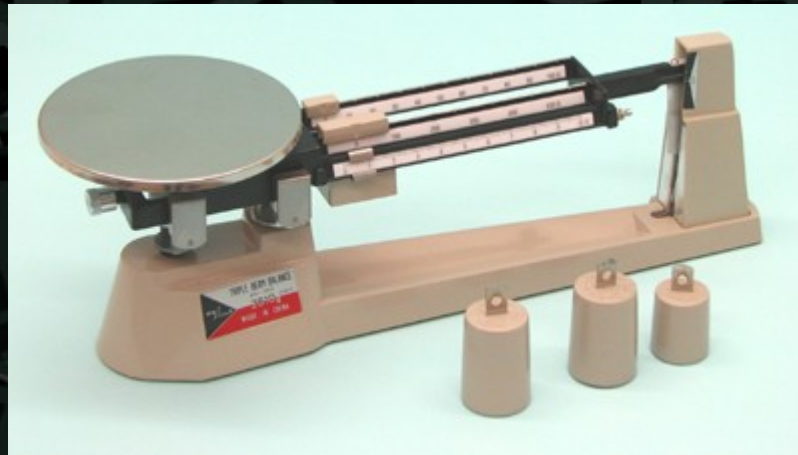
DENSITY = MASS/VOLUME

In other words... amount of stuff/amount of space

UNITS:  $\text{g/cm}^3$

So to get DENSITY, need to know mass and volume

How do I find out the mass?



# How do I find out the volume?

- Easy if it's a perfect sphere, or a cube...
- But WHAT IF it's something irregular?

# DUNK IT IN WATER!

- 1 mL of water occupies exactly  $1 \text{ cm}^3$   
=> at room temperature, anyway
- Place object in graduated cylinder, see how much water it displaces (i.e. how much water level changes)
- This equals the volume

# Archimedes and the Crown

